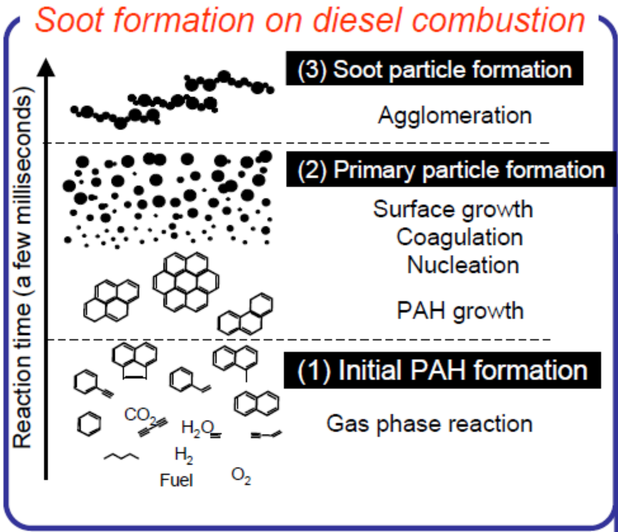
Chemical kinetic modeling of oxygenated fuels

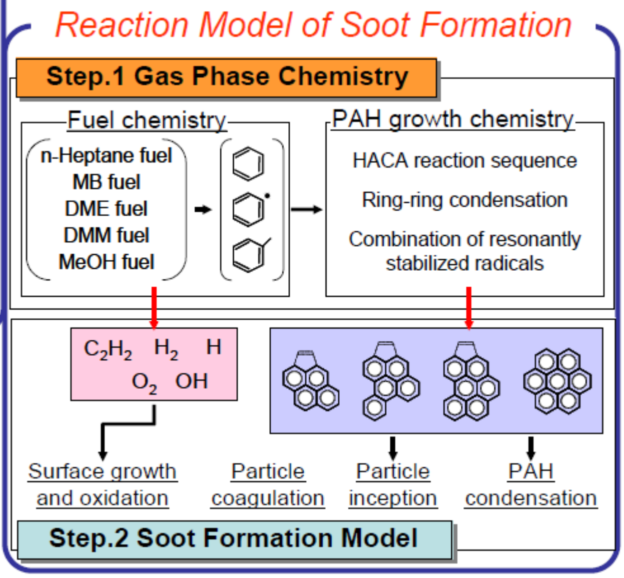


Oxygenated fuels

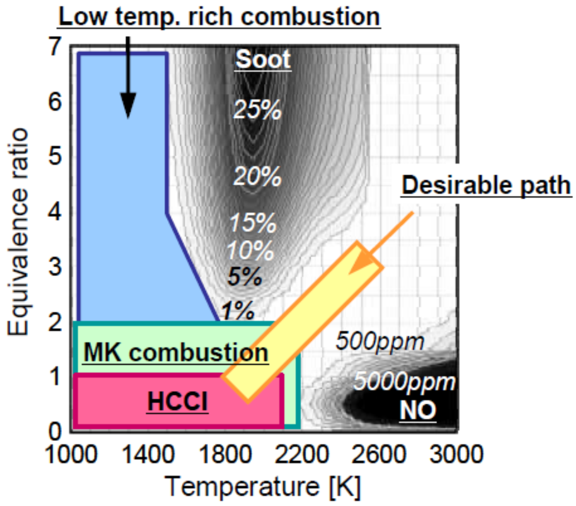
⇩

Soot suppression effect

Soot model { PAH particle formation / Initial PAH formation }

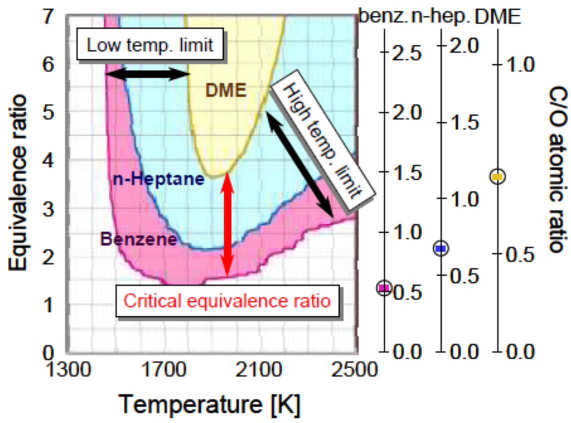


Chemical kinetic modeling of oxygenated fuels

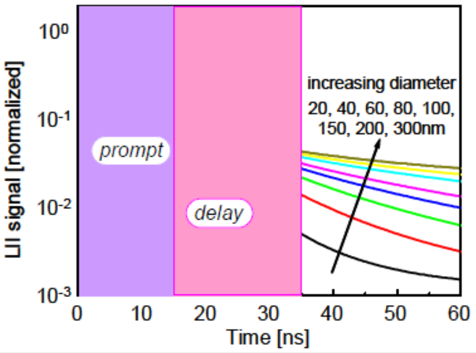


Comparison of Representative Diesel Combustion Methods on ϕ-T Diagram

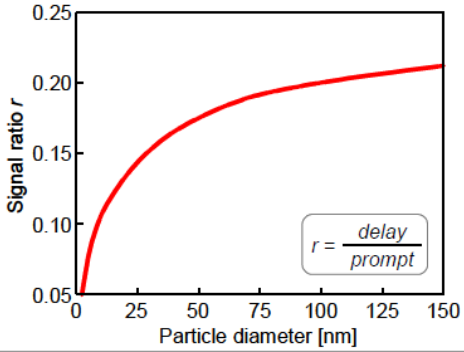
Variation of Soot Formation Limits among Different Type of Fuels on ϕ-T Diagram



Time-resolved LII



P_{amb}=4.1MPa, Laser fluence=1.67J/cm², T_{flame}=2200K



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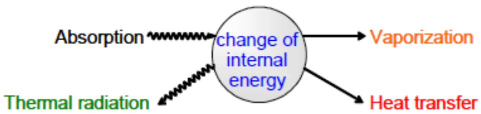


(Stefan Will, et al, APPLIED OPTICS, Vol.37, No.24, pp.5647-5658, 1998)

Energy balance equation

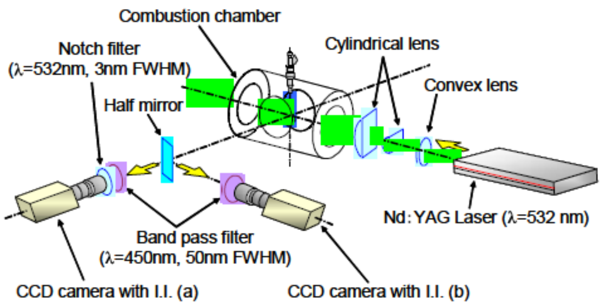
$$Q_{abs}\pi a^2 q_{(t)} - 4\pi a^2 (T - T_0)\Lambda - \frac{\Delta H_v}{W_s}\cdot\frac{dM}{dt} - q_{rad} - \frac{4}{3}\pi a^3 \rho_s C_s \frac{dT}{dt} = 0$$

absorbed laser energy | heat transfer loss | heat loss soot evaporation | heat loss thermal radiation | internal energy change

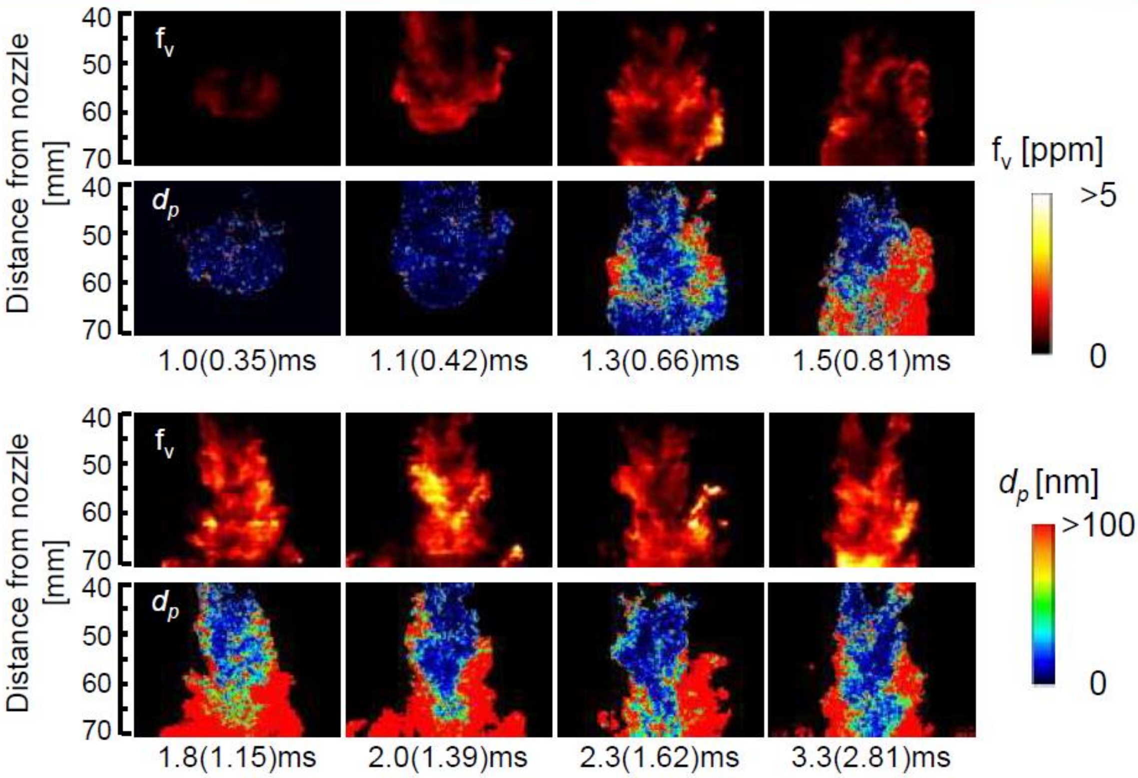
Mass conservation equation

$$\frac{dM}{dt} = 4\pi a^2 \rho_s \frac{da}{dt} = 4\pi a^2 \rho_v \sqrt{\frac{RT}{2W_v}}$$

(Stefan Will, et al, APPLIED OPTICS, Vol.37, No.24, pp.5647-5658, 1998)



f_v and d_p distribution for heptane (Δt_{inj}=4.0ms)



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